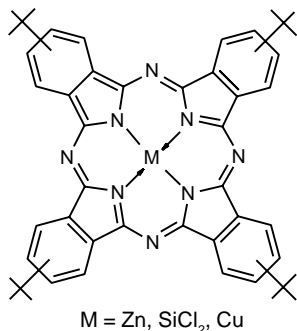


Functional Dyes

Near-Infrared Absorbers

During the past few years, we have introduced several new groups of functional dyes^{1,2} including infrared absorbers,^{3,4} materials for nonlinear optics, soluble phthalocyanine- and naphthalocyanine, and laser dyes.⁵ These materials have been used by high-tech companies and research laboratories for developing new, unconventional technologies⁶ such as optical recording, thermal writing, and laser printing.

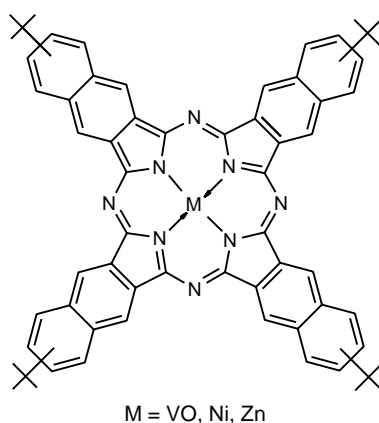
Recently, we have concentrated our efforts on two groups of near-infrared absorbers that absorb in the 670-810nm region. One absorber group includes highly photoresistant tetra-*tert*-butylphthalocyanine- and -naphthalocyanines; the second group includes photolabile carbocyanine dyes. Both groups of materials have very intense, narrow absorption bands with E values exceeding 200,000. These two complementary classes allow the user to cover a wide spectrum of applications.



43,099-4 Zinc 2,9,16,23-tetra-*tert*-butyl-29*H*,31*H*-phthalocyanine, $\lambda_{\max} = 677\text{nm}$

43,218-0 Silicon 2,9,16,23-tetra-*tert*-butyl-29*H*,31*H*-phthalocyaninedichloride, $\lambda_{\max} = 672\text{nm}$

42,316-5 Copper(II) 2,9,16,23-tetra-*tert*-butyl-29*H*,31*H*-phthalocyanine, $\lambda_{\max} = 672\text{nm}$

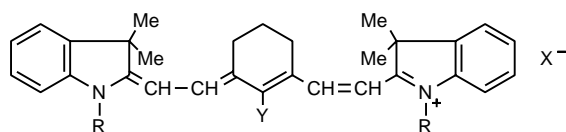


43,296-2 Vanadyl 2,11,20,29-tetra-*tert*-butyl-2,3-naphthalocyanine, $\lambda_{\max} = 808\text{nm}$

43,295-4 Nickel(II) 2,11,20,29-tetra-*tert*-butyl-2,3-naphthalocyanine, $\lambda_{\max} = 766\text{nm}$

43,221-0 Zinc 2,11,20,29-tetra-*tert*-butyl-2,3-naphthalocyanine, $\lambda_{\max} = 769\text{nm}$

42,527-3 2,11,20,29-Tetra-*tert*-butyl-2,3-naphthalocyanine, $\lambda_{\max} = 784\text{nm}$



42,531-1 IR-780 iodide, R = Pr; X⁻ = I⁻; Y = Cl; $\lambda_{\max} = 780\text{nm}$

42,413-7 IR-786 iodide, R = Me; X⁻ = I⁻; Y = Cl; $\lambda_{\max} = 775\text{nm}$

42,530-3 IR-780 perchlorate, R = Pr; X⁻ = ClO₄⁻; Y = Cl; $\lambda_{\max} = 780\text{nm}$

40,711-9 IR-786 perchlorate, R = Me; X⁻ = ClO₄⁻; Y = Cl; $\lambda_{\max} = 775\text{nm}$

42,598-2 IR-792 perchlorate, R = Pr; X⁻ = ClO₄⁻; Y = SPh; $\lambda_{\max} = 792\text{nm}$

42,745-4 IR-768 perchlorate, R = Pr; X⁻ = ClO₄⁻; Y = OPh; $\lambda_{\max} = 768\text{nm}$

References: (1) Zollinger, H. *Color Chemistry: Syntheses, Properties and Applications of Organic Dyes and Pigments*, 2nd rev. ed.; VCH: Weinheim, Germany, 1991. (Aldrich Cat. No. **Z23,335-8**) (2) *Phthalocyanines, Properties and Applications*; Leznoff, C.C.; Lever, A.B.P., Eds.; VCH: New York, 1993. (3) *Infrared Absorbing Dyes*; Matsuoka, M., Ed.; Plenum: New York, 1990. (Aldrich Cat. No. **Z23,583-0**) (4) Fabian, J. et al. *Chem. Rev.* **1992**, *92*, 1197. (5) Luk'yanetz, E.A. *Mol. Mat.* **1992**, *1*, 15. (6) *Nonlinear Optical Properties of Organic Molecules and Crystals*; Chémia, D.S.; Zyss, J., Eds.; Academic: Orlando, 1987, Vol. 1 and 2.

A Related Product

42,815-9 Silicon(IV) phthalocyanine bis(trihexylsilyloxyde), $\lambda_{\max} = 669\text{nm}$



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