

Battery Materials for the 21st Century

What will energy sources of the next century look like? The race is on to answer that question. Research is advancing with hundreds of papers advocating different potential solutions — some entail major departures from the present reliance on lead-acid, nickel-cobalt, or lithium ion batteries; others include only minor improvements; while yet others involve the synthesis of new substances as novel active battery materials.¹ For example:

- ◆ Phosphomolybdic acid as a precursor to hybrid organic-inorganic based molecular batteries with applications as cation-insertion electrodes.²
- ◆ Poly(3-octylthiophene) employed in multilayer polymer film electrodes.³
- ◆ Phosphotungstic acid solution as the electrolyte in a low cost, simple fuel cell, resistant to carbon monoxide poisoning.⁴

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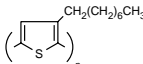




(1) Kawai, H. et al. *Chem. Mater.* **1998**, *10*, 3266. (2) Lira-Cantú, M.; Gómez-Romero, P.; *Chem. Mater.* **1998**, *10*, 698. (3) Kvarnstrom, C. et al. *Mater. Res. Soc. Symp. Proc.* **1997**, 451. (4) Hocevar, S. et al. *New Mater. Fuel Cell Mod. Battery Syst. II, Proc. Int. Symp., 2nd, 1997*, 297.



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Quality Materials for Research

Co_3O_4	20,311-4 Cobalt(II,III) oxide, 99.995%	5g, 25g
MnO_2	20,375-0 Manganese(IV) oxide, 99.99%	5g, 25g
MoS_2	23,484-2 Molybdenum(IV) sulfide, powder, < 2 μm , 99%	100g, 500g
NiO	20,388-2 Nickel(II) oxide, 99.99%	20g, 100g
$\text{NiO}_x \cdot x\text{H}_2\text{O}$	36,719-2 Nickel peroxide hydrate	10g, 50g
$\text{H}_3\text{PO}_4 \cdot 12\text{MoO}_3 \cdot x\text{H}_2\text{O}$	43,140-0 Phosphomolybdic acid hydrate, <i>ReagentPlus</i> TM , 99.99+ %	10g, 50g
$\text{H}_3\text{PO}_4 \cdot 12\text{WO}_3 \cdot x\text{H}_2\text{O}$	45,597-0 Phosphotungstic acid hydrate, 99.995%	10g, 50g
	44,571-1 Poly(3-octylthiophene-2,5-diyl), regioregular	1g
	48,255-2 Polypyrrole, 5 wt. % solution in water	100mL
$\text{H}_4[\text{Si}_4(\text{W}_3\text{O}_9)_4] \cdot x\text{H}_2\text{O}$	38,334-1 Silicotungstic acid hydrate, 99.9+ %	10g, 50g
SnO	24,464-3 Tin(II) oxide, powder, 10 μm , 99+ %	100g, 500g
WO_3	20,478-1 Tungsten(VI) oxide, 99.995%	50g, 250g
V_2O_3	46,374-4 Vanadium(III) oxide, 99.99%	5g, 25g
V_2O_5	20,485-4 Vanadium(V) oxide, 99.99%	5g, 25g
$\text{ZrO}_2/\text{Y}_2\text{O}_3$	46,420-1 Zirconium(IV) oxide, yttria stabilized (~5.3% Y_2O_3), submicron powder, 99.5%	100g, 500g
$\text{Li}[\text{N}(\text{CF}_3\text{SO}_2)_2]$	44,950-4 N-Lithiotrifluoromethanesulfonimide, 97%	10g, 50g
LiCoO_2	44,270-4 Lithium cobalt(III) oxide, 99.8%	25g, 100g
LiPF_6	45,022-7 Lithium hexafluorophosphate, 99.99+ %	5g, 25g
LiMn_2O_4	48,227-7 Lithium manganese(III,IV) oxide, electrochemical grade	25g, 100g
LiClO_4	43,156-7 Lithium perchlorate, <i>ReagentPlus</i> TM , 99.99%	50g, 250g
LiAlCl_4	42,947-3 Lithium tetrachloroaluminate, anhydrous, powder, 99.99%	1g, 10g
	D15,292-7 Dimethyl carbonate, 99%	100g, 500g
	E2,625-8 Ethylene carbonate, 98%	100g, 500g
SO_2	29,569-8 Sulfur dioxide, 99.9+ %	454g
SOCl_2	44,728-5 Thionyl chloride, low iron, 99.5%, (Fe < 5ppm)	100mL, 1L
	V260-7 Vinylene carbonate, 97%	5g, 25g



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