



Product Information

D-(+)-Glucose solution Cell Culture Tested

Product Number **G 8769**
Store at Room Temperature

Product Description

Molecular Formula: $C_6H_{12}O_6$

Molecular Weight: 180.2

CAS Number: 50-99-7

pH: 3 – 4 (25 °C)

The density of an aqueous 40% solution (w/v) at 17.5 °C is 1.149 g/ml.²

This product is a 45% (w/v) solution of glucose. Based on a 45% (w/v) concentration and a molecular weight for glucose of 180 Da, the solution is 2.5 M glucose.

This product has been tested with cell lines to verify the product is not cytotoxic. The growth promoting capacity of this cell culture tested glucose solution was assessed at 10 ml/L in medium using appropriate cell lines.

Glucose is a main source of energy for living organisms. Glucose occurs naturally in the free state in fruits and other parts of plants. Glucose is combined into glucosides, disaccharides, oligosaccharides, the polysaccharides (cellulose and starch), and glycogen.

Glucose is a mixture of α - and β -anomers, primarily the α -anomer. The optical rotation of the α -anomer is +112.2° (c = 10% in water, 20 °C) and the β -anomer is +18.7° (c = 10% in water, 20 °C). When D-glucose is dissolved in water, the optical rotation gradually changes (mutarotates) with time and approaches a final equilibrium value of +52.7° (c = 10%, 20 °C) due to the formation of an equilibrium mixture consisting of approximately one-third α - and two-thirds β -D-glucose.²

Precautions and Disclaimer

For Laboratory Use Only. Not for drug, household or other uses.

Storage/Stability

This solution has been sterilized by autoclaving.

References

1. Biochemistry, 2nd ed., Lehninger, A. L., ed., Worth Publishers, Inc. (New York, NY: 1975), p. 253.
2. The Merck Index, 13th Ed., Entry# 4472.
3. Puget, K., and Michelson, A. M., Microestimation of glucose and glucose oxidase. *Biochimie*, **58**, 757-758 (1976).

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