

Product Information

Deoxyribonuclease I, bovine recombinant, expressed in *Pichia pastoris* buffered aqueous glycerol solution

Catalog Number **D5319**
Storage Temperature $-20\text{ }^{\circ}\text{C}$

CAS RN 9003-98-9
EC 3.1.21.1
Synonyms: DNase I,
Deoxyribonuclease 5'-Oligonucleotidohydrolase

Product Description

Deoxyribonuclease I (DNase I) is found in most cells and tissues. In mammals, the pancreas is one of the best sources for the enzyme. Pancreatic DNase I was the first isolated DNase.

DNase I is an endonuclease that acts on phosphodiester bonds adjacent to pyrimidines to produce polynucleotides with terminal 5'-phosphates. A tetranucleotide is the smallest average digestion product. DNase I hydrolyzes single- and double-stranded DNA. In the presence of Mg^{2+} ions, DNase I attacks each strand of DNA independently and the cleavage sites are random. If Mn^{2+} ions are present, both DNA strands are cleaved at approximately the same site.¹ When chromatin DNA is digested, the reaction rate is restricted by the association of DNA with histones.¹

DNase I can be used to remove DNA from protein and nucleic acid samples, and to nick DNA as a first step to incorporate labeled bases into DNA.

Activators:

DNase I has an absolute requirement for divalent metal cations. Mg^{2+} is the most commonly used divalent cation.² Mn^{2+} , Ca^{2+} , Co^{2+} , and Zn^{2+} will also activate DNase I.^{2,3} A concentration of 5 mM Ca^{+2} will stabilize DNase I against proteolytic digestion. 0.1 mM Ca^{+2} is needed to reduce the rate of inactivation by one-half.⁴

Inhibitors:

2-Mercaptoethanol (the reduced enzyme is inactive, but can be reactivated in the presence of Ca^{2+} or Mg^{2+} ions),³ chelators (e.g. EDTA); sodium dodecyl sulfate (SDS),⁵ actin.⁷ There is no general inhibitor specific for DNase I.² Citrate inhibits Mg^{2+} -activated DNase I, but not Mn^{2+} -activated DNase I.

Optimal pH:

The optimal pH of DNase I activity is dependent on the divalent ion present. In the presence of both Mg^{2+} and Ca^{2+} , the optimal pH is between 7–8, while in the absence of Ca^{2+} , the optimal pH is between 5.5–6.0.⁶

Extinction Coefficient: $E_{280}^{1\%} = 11.1$

This recombinant bovine DNase I is a glycoprotein, produced without the addition of any animal-derived materials. It is supplied as a **solution** in 4 mg/ml glycine, pH 5.0, 5 mM calcium acetate, and 50% glycerol.

Molecular mass: ~39 kDa

Specific activity: $\geq 5,000$ units/mg protein

Unit definition: One unit will produce a change in A_{260} of 0.001 per minute per ml at pH 5.0 at 25 °C using DNA, Type I or III, as the substrate. This enzyme assay reaction is performed in 83 mM acetate buffer, pH 5.0, at 25 °C, containing 4.2 mM Mg^{2+} , in a 3 ml reaction.

Impurities:

Protease – None Detected
RNase – None detected

Precautions and Disclaimer

This product is for R&D use only, not for drug, household, or other uses. Please consult the Safety Data Sheet for information regarding hazards and safe handling practices.

Storage/Stability

This product retains activity for at least two years when stored at $-20\text{ }^{\circ}\text{C}$.

References

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2. Weir, A.F., in *Enzymes of Molecular Biology*, Vol. 16 of *Methods in Molecular Biology*, Burrell, M.M., ed. (Humana Press, 1993), Ch. 2, 2-16
3. Price, P.A., *et al.*, Effect of divalent cations on the reduction and re-formation of the disulfide bonds of deoxyribonuclease. *J. Biol Chem.*, **244**, 929-932 (1969).
4. Price, P.A., *et al.*, Properties of chromatographically purified bovine pancreatic deoxyribonuclease. *J. Biol Chem.*, **244**, 917-923 (1969).
5. Liao, T.-H., Reversible inactivation of pancreatic deoxyribonuclease A by sodium dodecyl sulfate. Removal of COOH-terminal residues from the denatured protein by carboxypeptidase A. *J. Biol. Chem.*, **250**, 3831-3836 (1975).
6. Love, J.D., and Hewitt, R.R., The relationship between human serum and human pancreatic DNase I. *J. Biol. Chem.*, **254**, 12588-12594 (1979).

DT,RC,GCY,LS,RBG,MAM 07/15-1