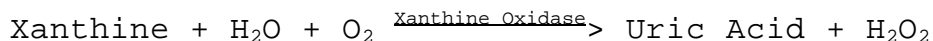


**Enzymatic Assay of XANTHINE OXIDASE
(EC 1.1.3.22)**

PRINCIPLE:



CONDITIONS: T = 25°C, pH = 7.5, A_{290nm}, Light Path = 1 cm

METHOD: Continuous Spectrophotometric Rate Determination

REAGENTS:

- A. 50 mM Potassium Phosphate Buffer, pH 7.5 at 25°C
(Prepare 200 ml in deionized water using Potassium Phosphate, Monobasic, Anhydrous, Sigma Prod. No. P-5379. Adjust to pH 7.5 at 25°C with 1 M KOH.)
- B. 0.15 mM Xanthine Solution
(Prepare 100 ml by initially dissolving Xanthine, Sigma Prod. No. X-0626, in a minimal volume of NaOH. Add approximately 90 ml of deionized water. Adjust to pH 7.5 at 25°C with either 1 M NaOH or 1 M HCl. Dilute to a final volume of 100 ml. **PREPARE FRESH.**)
- C. Xanthine Oxidase Enzyme Solution
(Immediately before use, prepare a solution containing 0.1 - 0.2 unit/ml of Xanthine Oxidase in cold Reagent A.)

PROCEDURE:

Pipette (in milliliters) the following reagents into suitable quartz cuvettes:

| | <u>Test</u> | <u>Blank</u> |
|----------------------|-------------|--------------|
| Reagent A (Buffer) | 1.90 | 1.90 |
| Reagent B (Xanthine) | 1.00 | 1.00 |
| Deionized Water | ----- | 0.10 |

**Enzymatic Assay of XANTHINE OXIDASE
(EC 1.1.3.22)**

PROCEDURE: (continued)

Mix by inversion and equilibrate to 25°C. Monitor the $A_{290\text{nm}}$ until constant, using a suitably thermostatted spectrophotometer. Then add:

| | <u>Test</u> | <u>Blank</u> |
|-----------------------------|-------------|--------------|
| Reagent C (Enzyme Solution) | 0.10 | ----- |

Immediately mix by inversion and record the increase in $A_{290\text{nm}}$ for approximately 5 minutes. Obtain the $r A_{290\text{nm}}$ /minute using the maximum linear rate for both the Test and Blank.

CALCULATIONS:

$$\text{Units/ml enzyme} = \frac{(r A_{290\text{nm}}/\text{min Test} - r A_{290\text{nm}}/\text{min Blank})(3)(\text{df})}{(12.2)(0.1)}$$

3 = Total volume (in milliliters) of assay

df = Dilution factor

12.2 = Millimolar extinction coefficient of Uric Acid
at 290 nm

0.1 = Volume (in milliliter) of enzyme used

$$\text{Units/mg solid} = \frac{\text{units/ml enzyme}}{\text{mg solid/ml enzyme}}$$

$$\text{Units/mg protein} = \frac{\text{units/ml enzyme}}{\text{mg protein/ml enzyme}}$$

UNIT DEFINITION:

One unit will convert 1.0 μmole of xanthine to uric acid per minute at pH 7.5 at 25°C.

FINAL ASSAY CONCENTRATION:

In a 3.00 ml reaction mix, the final concentrations are 33 mM potassium phosphate, 0.050 mM xanthine and 0.01 - 0.02 unit xanthine oxidase.

**Enzymatic Assay of XANTHINE OXIDASE
(EC 1.1.3.22)**

REFERENCE:

Bergmeyer, H.U., Gawehn, K. and Grassl, M. (1974) in *Methods of Enzymatic Analysis* (Bergmeyer, H.U. ed.) Second Edition, Volume I, 521-522, Academic Press Inc., New York, NY

NOTES:

1. This assay is based on the cited reference.
2. Where Sigma Product or Stock numbers are specified, equivalent reagents may be substituted.

This procedure is for informational purposes. For a current copy of Sigma's quality control procedure contact our Technical Service Department.